

The *trans*-Difluorotetramminechromium(III) Complex and Other Novel Compounds of the *trans*-Tetramminechromium(III) Series

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THE following would seem to be the first *trans*-tetramminechromium(III) compounds to be described. On the basis of the reaction between difluoro-tetrapyridinechromium(III) iodide^{1,2} and liquid ammonia in an autoclave at 100° giving a mixture of *cis*- and *trans*-difluorotetramminechromium(III) iodide, we have prepared the pure *trans*-difluoro-complex and several other crystalline salts of the *trans*-tetrammine series.

trans-[Cr(NH₃)₄F₂]I (slightly brownish red, *cis*-complex violet red) was isolated as the least soluble component of the above-*cis-trans*-iodide mixture by a recrystallisation from water (acidified with a few drops of HCl and at 70°) under addition of NaI. Reaction between *trans*-[Cr(NH₃)₄F₂]ClO₄ and 70% HClO₄ (1 hr., room temperature) gave [Cr(NH₃)₄F(H₂O)](ClO₄)₂ (slightly brownish red). With aqueous HBr (prepared by saturation at 1 atm. and -10°) at room temperature in an autoclave overnight [Cr(NH₃)₄F₂]ClO₄ gave *trans*-[Cr(NH₃)₄Br₂]Br (green) which could be aquated

to give *trans*-[Cr(NH₃)₄Br(H₂O)]Br₂ (olive green) and *trans*-[Cr(NH₃)₄(H₂O)₂]Br₃ (brick red). The diaquo-complex was isolated through the *trans*-[Cr(NH₃)₄(OH)(H₂O)]Br₂ (pink).

On analogous treatment, but with HCl, *trans*-[Cr(NH₃)₄F₂]ClO₄ gave a mixture of the dichloro- and the chloro-aquo-complexes. Slightly acidified water at 30° extracted the chloro-aquo-complex, which could then be isolated as *trans*-[Cr(NH₃)₄Cl(H₂O)]Cl₂ (reddish brown) by precipitation with concentrated HCl at 0°. The residue from the extraction was purified as *trans*-[Cr(NH₃)₄Cl₂]Cl (greyish green) by rapid dissolution in water (30°) and reprecipitation with conc. HCl at 0°.

The configurations of the tetrammine complexes were unambiguously established by comparison of their ligand field spectra with those of the corresponding bis(ethylenediamine) complexes. No *trans-cis*-isomerisation reactions were observed.

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